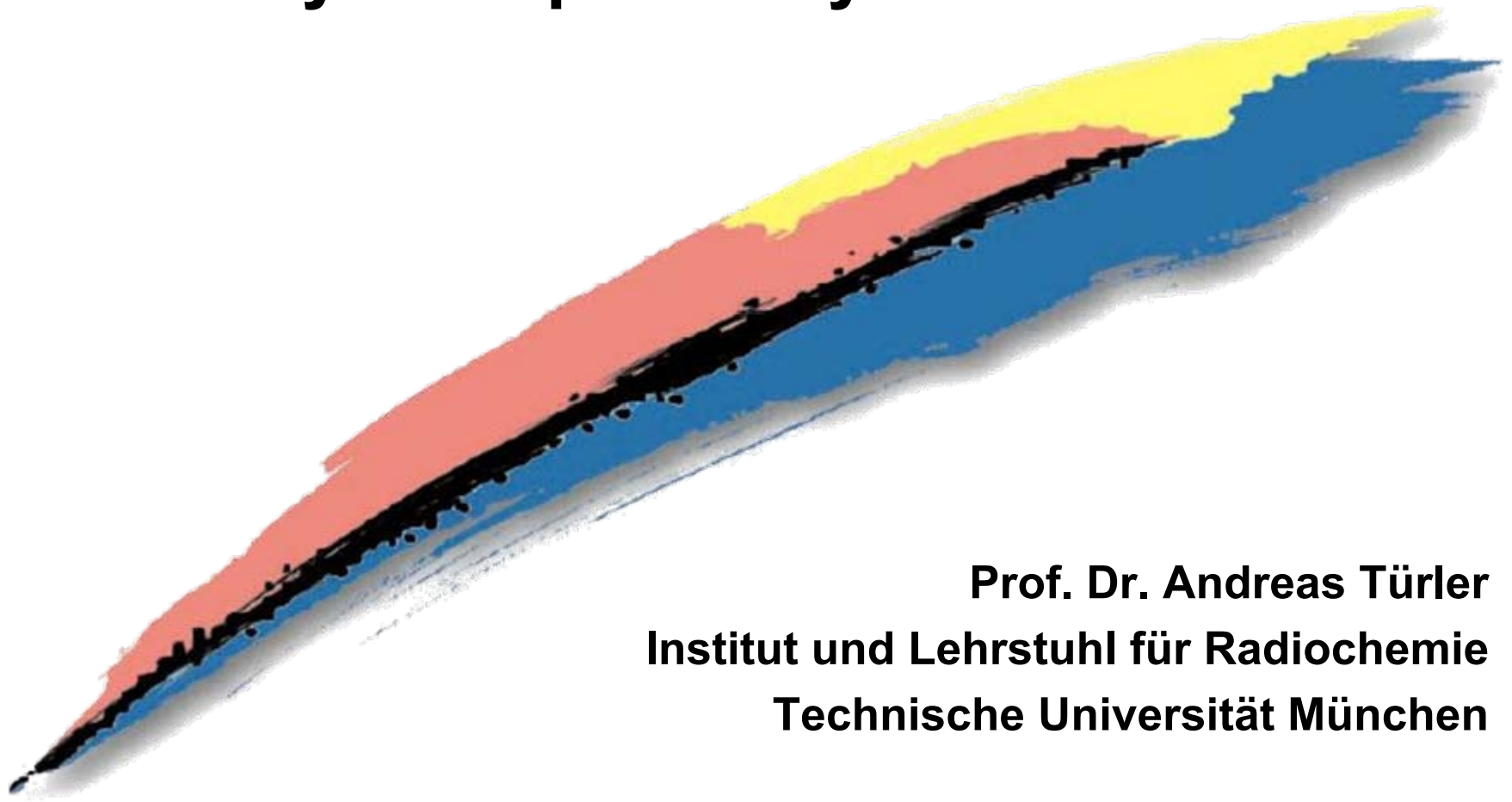


# Chemistry of Superheavy Elements



**Prof. Dr. Andreas Türler**  
**Institut und Lehrstuhl für Radiochemie**  
**Technische Universität München**

Logo of NRC6 conference Aachen

# Acknowledgements



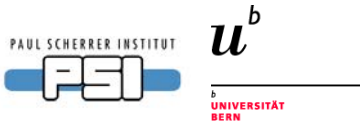
A.B. Yakushev, J. Dvorak, A. Semchenkov et al.,



J.V. Kratz, N. Trautmann et al.,



M. Schädel, Ch.E. Düllmann, V.G. Pershina et al.,



H.W. Gäggeler, R. Eichler et al.,



K.E. Gregorich et al.,



Y. Oganessian, A. Popeko, A. Yeremin et al.,

# Where ends the periodic table of the elements?

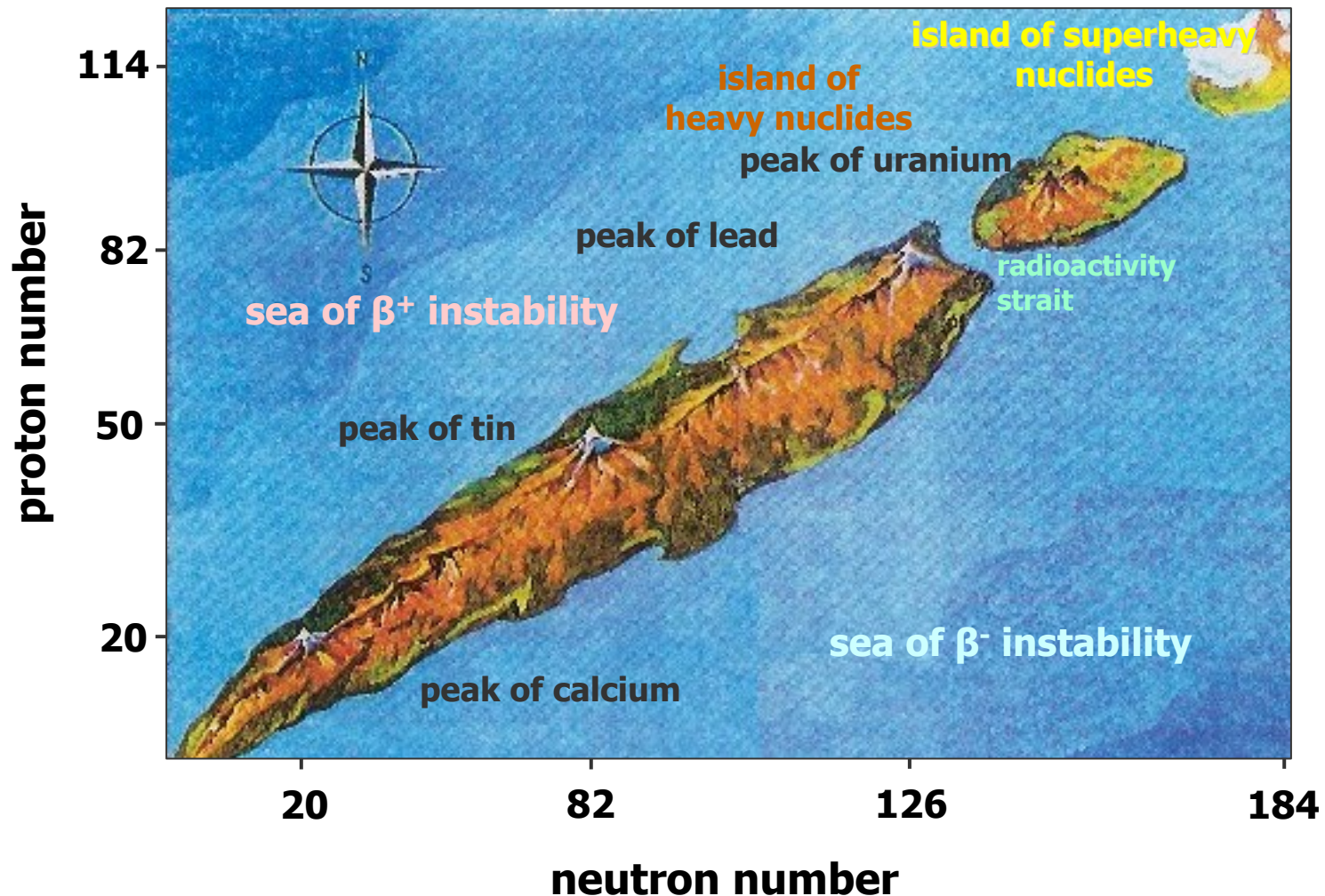
|          |          |              |           |           |           |           |           |           |           |           |          |          |          |          |          |          |               |
|----------|----------|--------------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|----------|----------|----------|----------|----------|----------|---------------|
| 1<br>H   | 2        |              |           |           |           |           |           |           |           |           |          | 13       | 14       | 15       | 16       | 17       | 18<br>2<br>He |
| 3<br>Li  | 4<br>Be  |              |           |           |           |           |           |           |           |           |          | 5<br>B   | 6<br>C   | 7<br>N   | 8<br>O   | 9<br>F   | 10<br>Ne      |
| 11<br>Na | 12<br>Mg | 3            | 4         | 5         | 6         | 7         | 8         | 9         | 10        | 11        | 12       | 13<br>Al | 14<br>Si | 15<br>P  | 16<br>S  | 17<br>Cl | 18<br>Ar      |
| 19<br>K  | 20<br>Ca | 21<br>Sc     | 22<br>Ti  | 23<br>V   | 24<br>Cr  | 25<br>Mn  | 26<br>Fe  | 27<br>Co  | 28<br>Ni  | 29<br>Cu  | 30<br>Zn | 31<br>Ga | 32<br>Ge | 33<br>As | 34<br>Se | 35<br>Br | 36<br>Kr      |
| 37<br>Rb | 38<br>Sr | 39<br>Y      | 40<br>Zr  | 41<br>Nb  | 42<br>Mo  | 43<br>Tc  | 44<br>Ru  | 45<br>Rh  | 46<br>Pd  | 47<br>Ag  | 48<br>Cd | 49<br>In | 50<br>Sn | 51<br>Sb | 52<br>Te | 53<br>I  | 54<br>Xe      |
| 55<br>Cs | 56<br>Ba | 57-71<br>La  | 72<br>Hf  | 73<br>Ta  | 74<br>W   | 75<br>Re  | 76<br>Os  | 77<br>Ir  | 78<br>Pt  | 79<br>Au  | 80<br>Hg | 81<br>Tl | 82<br>Pb | 83<br>Bi | 84<br>Po | 85<br>At | 86<br>Rn      |
| 87<br>Fr | 88<br>Ra | 89-103<br>Ac | 104<br>Rf | 105<br>Db | 106<br>Sg | 107<br>Bh | 108<br>Hs | 109<br>Mt | 110<br>Ds | 111<br>Rg | 112      | 113      | 114      | 115      | 116      |          | 118           |

Lanthanides

|          |          |          |          |          |          |          |          |          |          |          |           |           |           |           |
|----------|----------|----------|----------|----------|----------|----------|----------|----------|----------|----------|-----------|-----------|-----------|-----------|
| 57<br>La | 58<br>Ce | 59<br>Pr | 60<br>Nd | 61<br>Pm | 62<br>Sm | 63<br>Eu | 64<br>Gd | 65<br>Tb | 66<br>Dy | 67<br>Ho | 68<br>Er  | 69<br>Tm  | 70<br>Yb  | 71<br>Lu  |
| 89<br>Ac | 90<br>Th | 91<br>Pa | 92<br>U  | 93<br>Np | 94<br>Pu | 95<br>Am | 96<br>Cm | 97<br>Bk | 98<br>Cf | 99<br>Es | 100<br>Fm | 101<br>Md | 102<br>No | 103<br>Lr |

Actinides

# The island of superheavy nuclides!



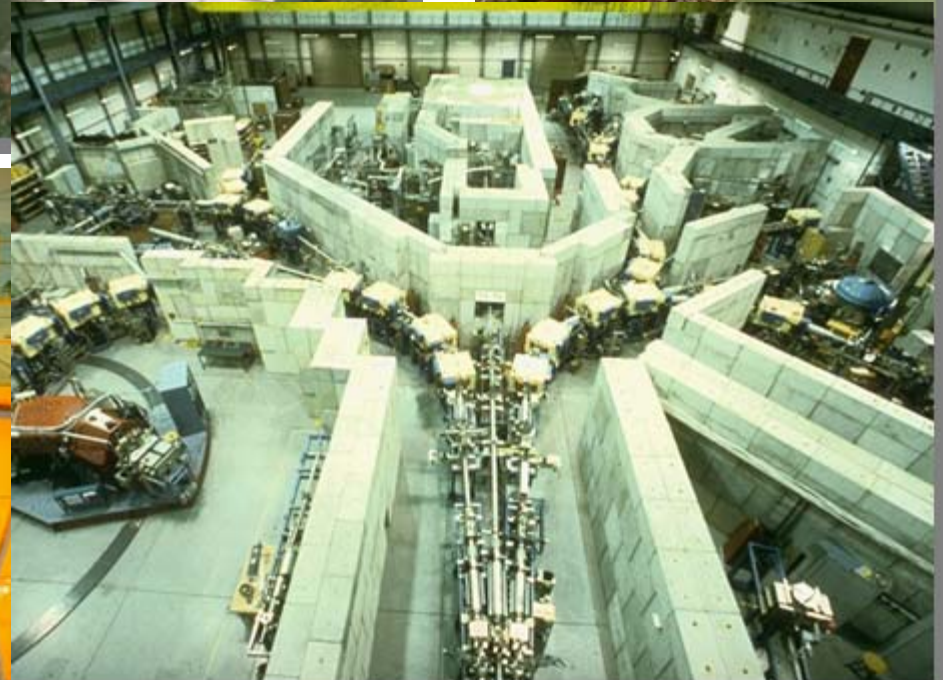
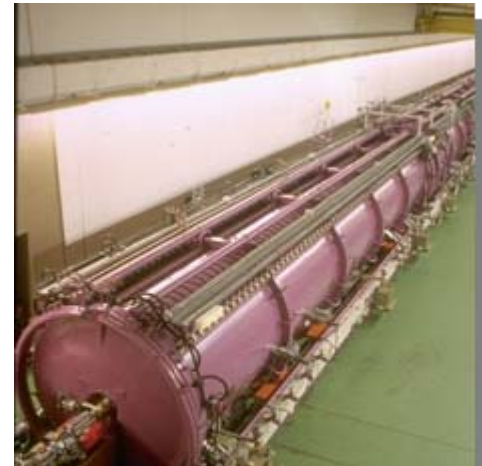
# Superheavy Element Synthesis

$1 \times 10^{17}$     $2 \times 10^{18}$

$5 \times 10^8$

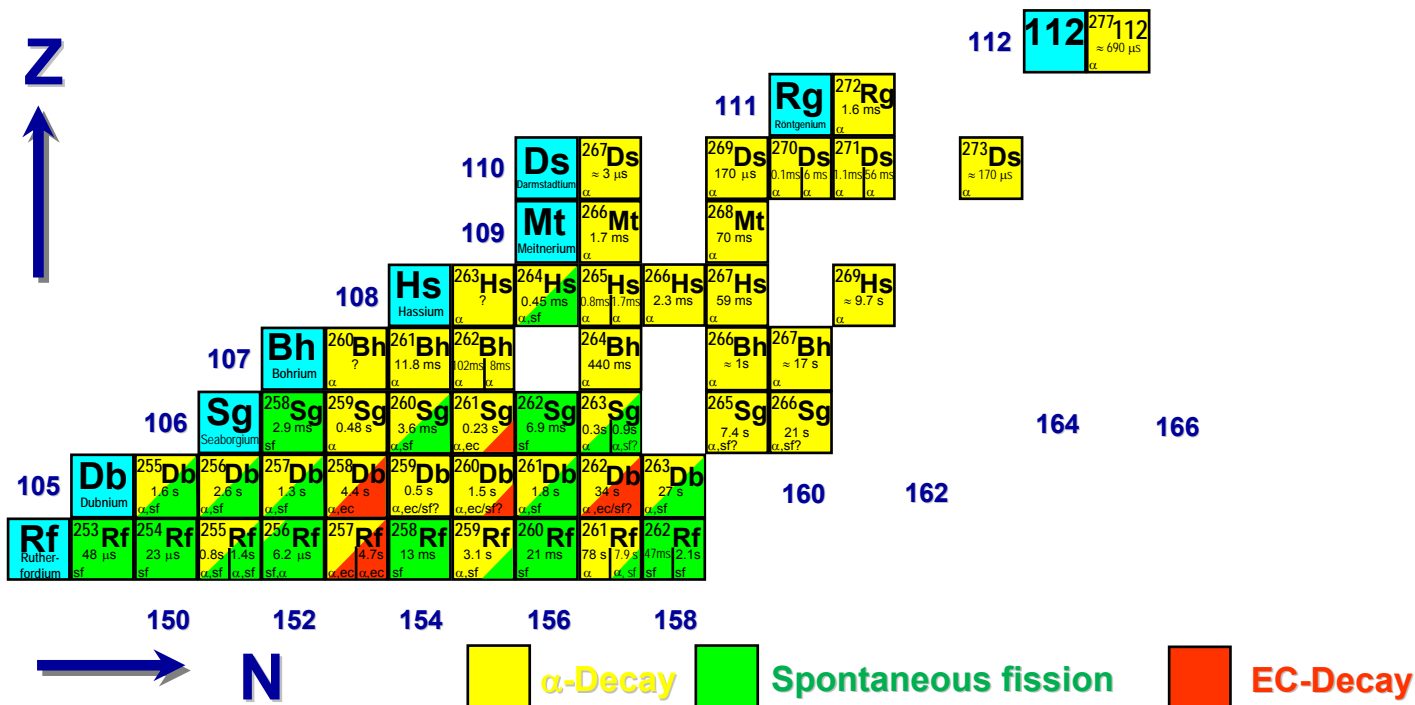
1

# GSI UNiversal Linear ACcelerator



# chart of nuclides (1998)

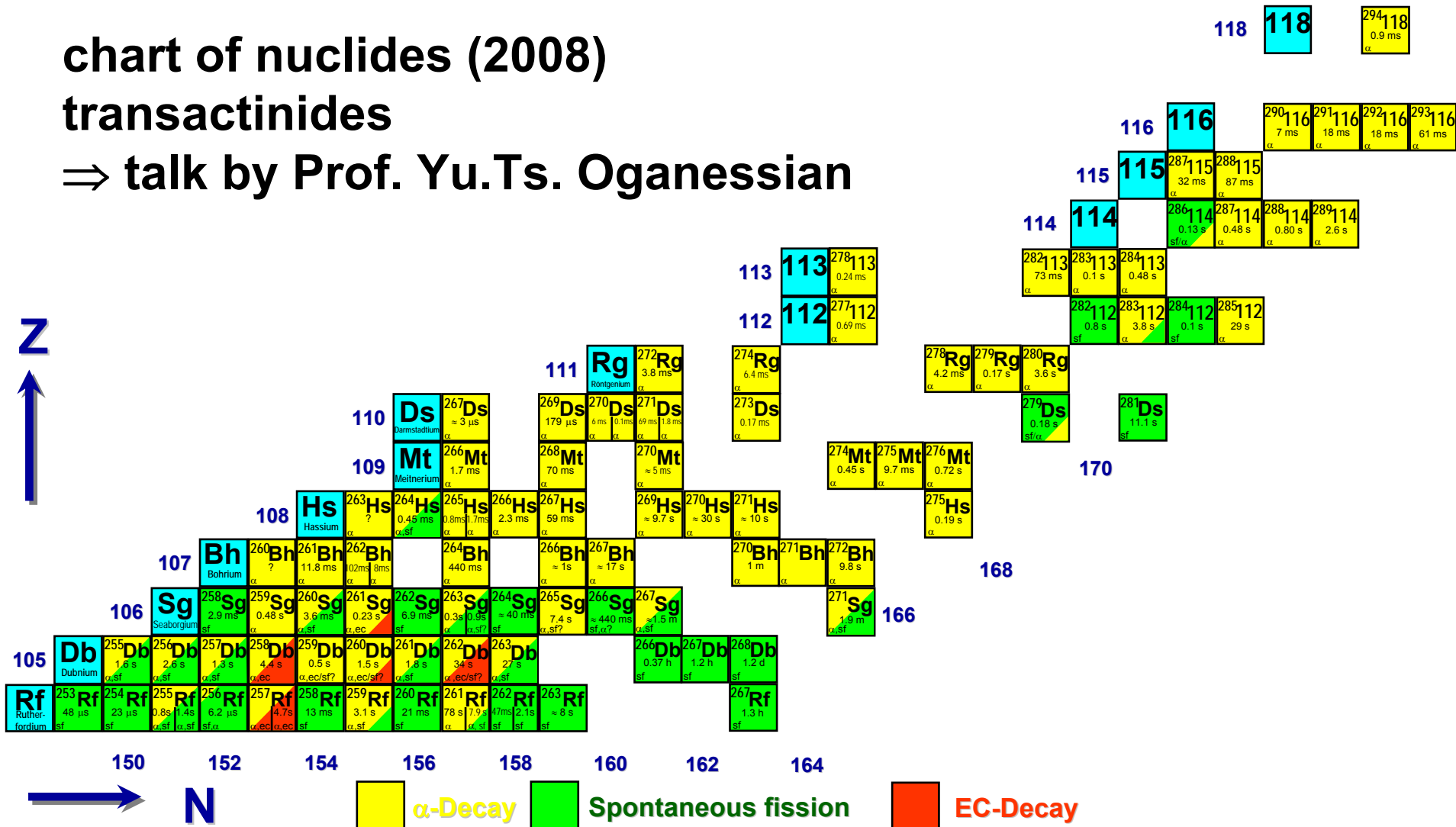
## transactinides



# chart of nuclides (2008)

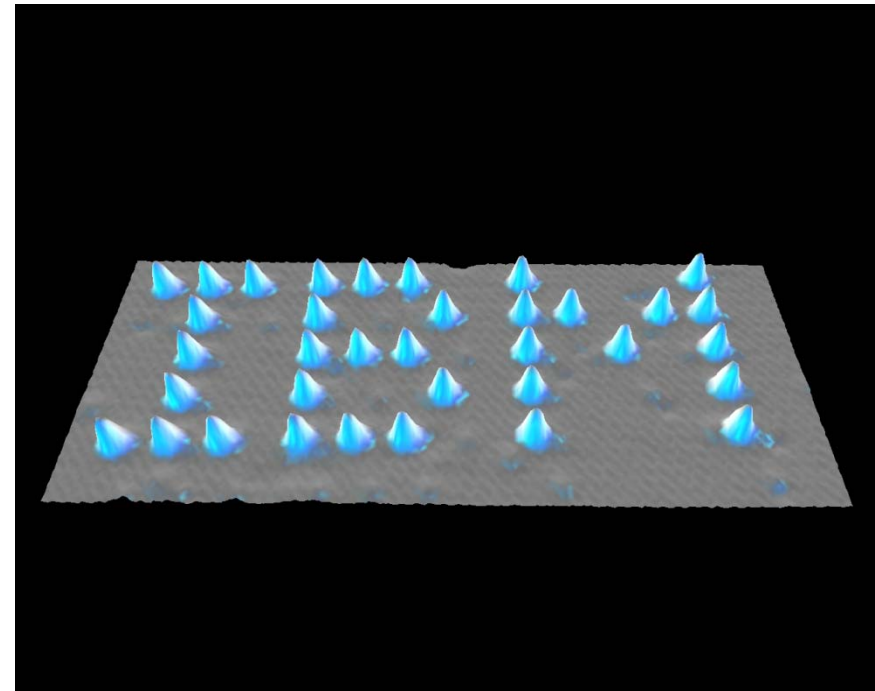
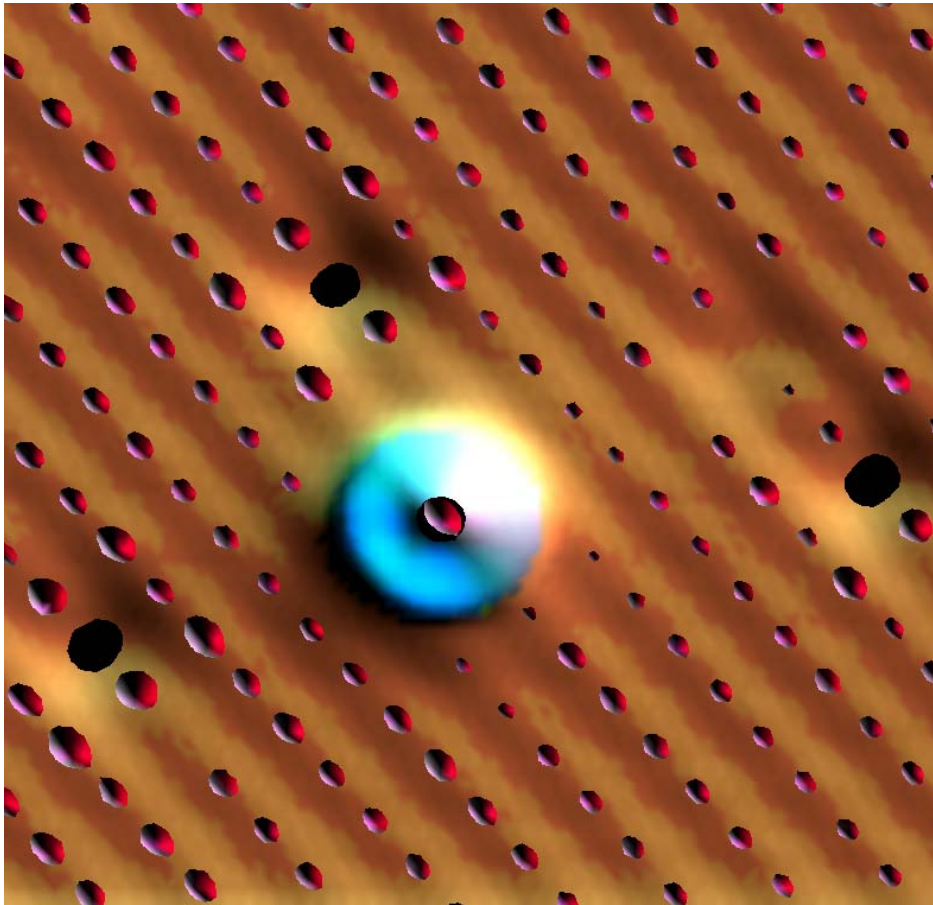
## transactinides

⇒ talk by Prof. Yu.Ts. Oganessian



# Manipulation of single atoms

## STM image of a Xe atom adsorbed on a Ni surface (110)



<http://www.almaden.ibm.com/vis/stm/gallery.html>

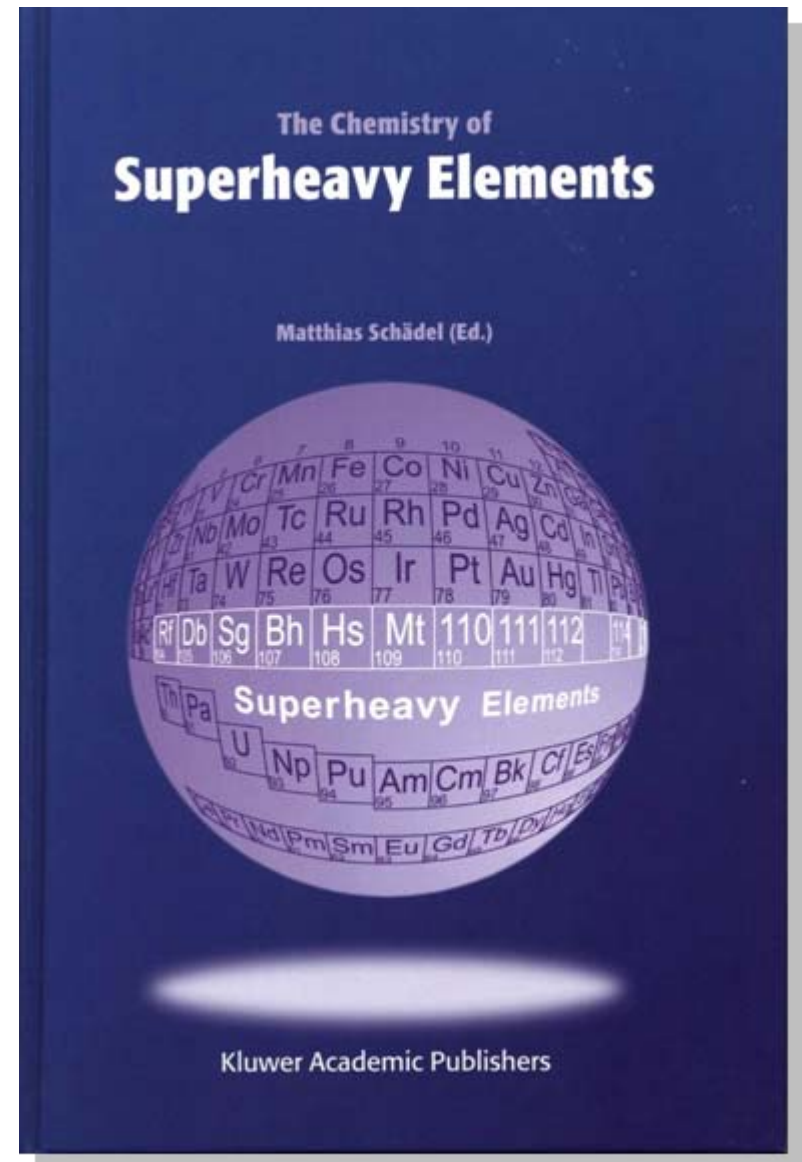
## Chemistry of “one (exotic) atom at the time”?



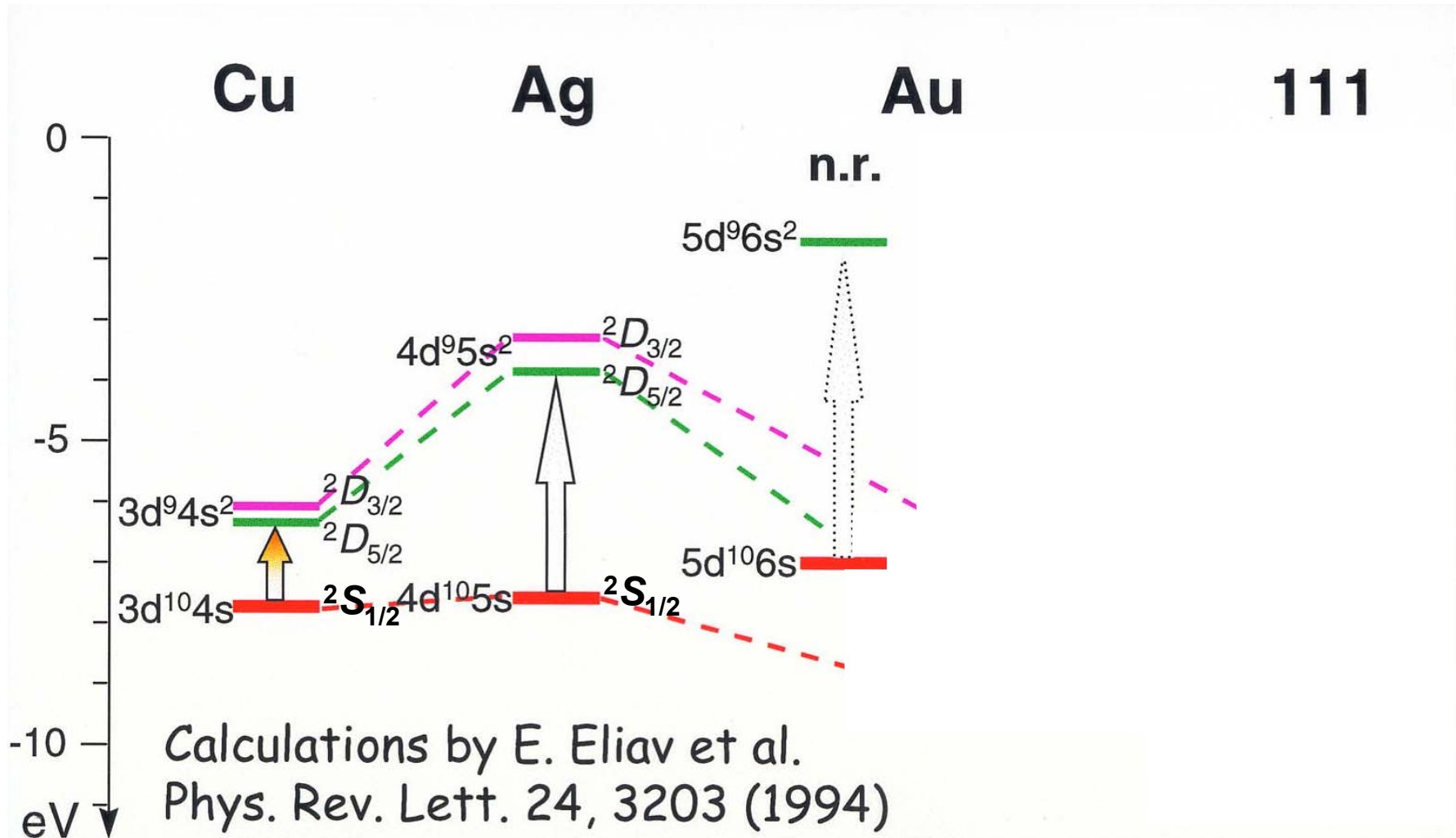


**Current status of superheavy  
element chemistry:**

**Read the book!**



# Relativistic effects: the color of gold!



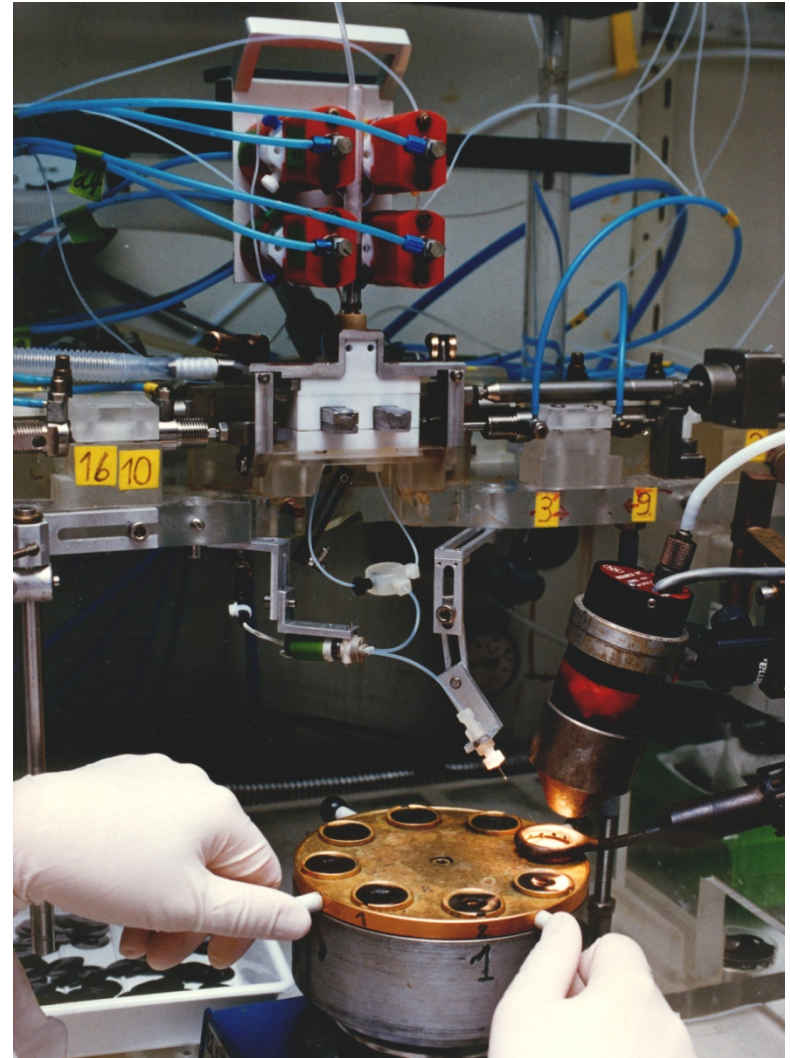
# Aqueous chemistry of Dubnium ( $Z=105$ )

Investigated with the  
**Automated Rapid Chemistry Apparatus**  
**ARCA**

J.V. Kratz & M. Schädel et al.,

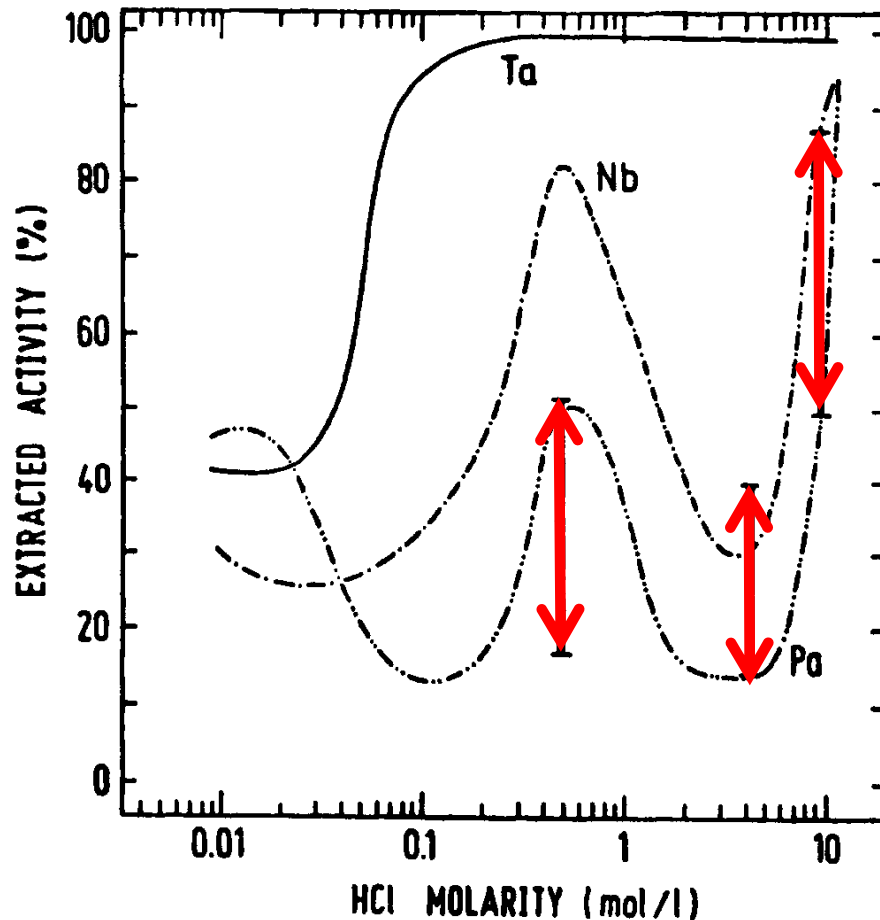
JOHANNES  
GUTENBERG  
UNIVERSITÄT  
MAINZ

GSII



# Non Ta-like behavior of Dubnium

H.P. Zimmermann et al., RCA 60, 11 (1993)



Percentage extracted activity of Nb, Ta, and Pa tracers (curves) as a function of HCl concentration in the System TiOA-HCl/0.03 M HF. The bold bars encompass the upper and lower limits deduced for the Db extraction from the elution positions in the chromatography experiments. The complete extraction of Db into TiOA from 12 M HCl / 0.02 M HF is not indicated for clarity.

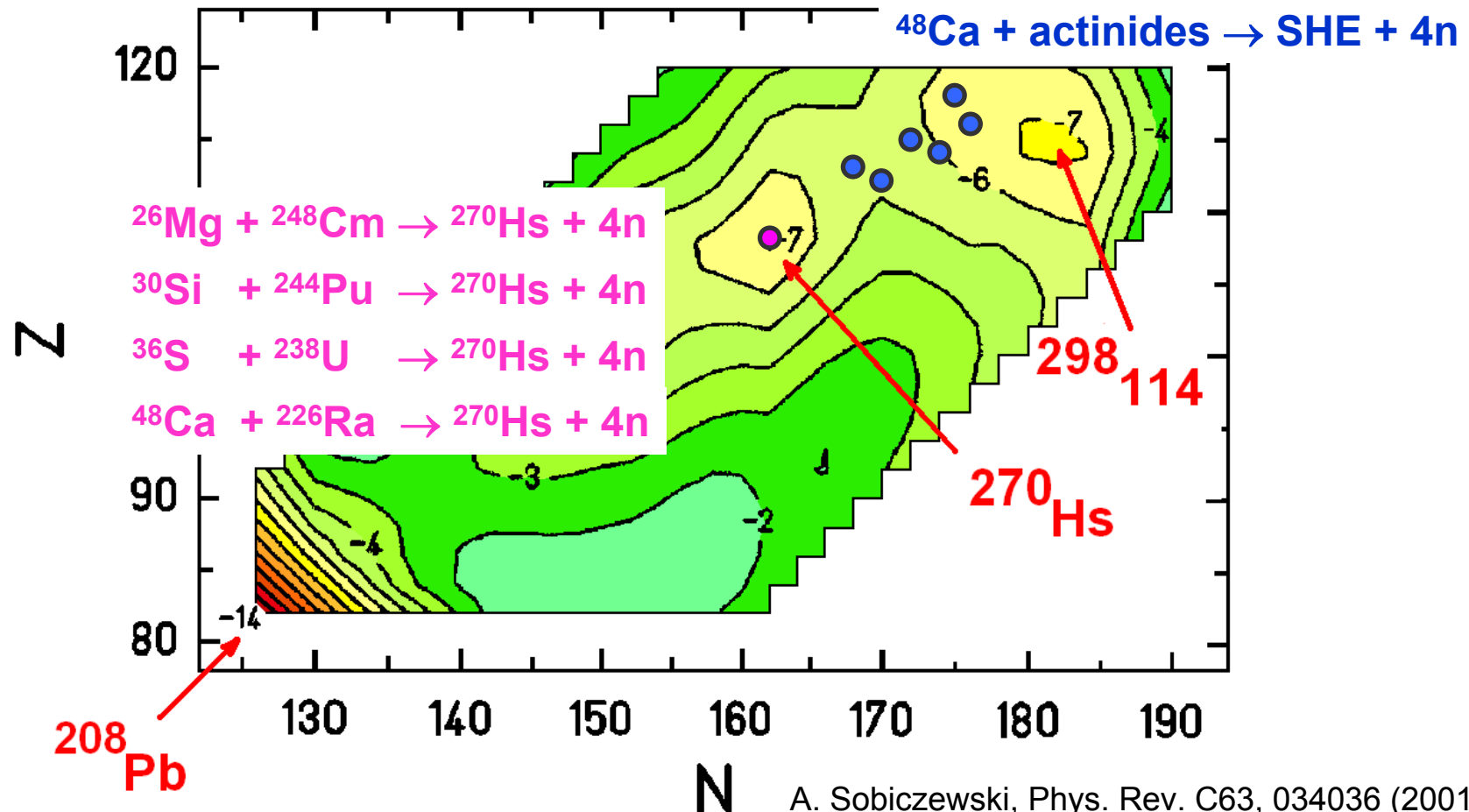
# Hassium (Z=108)



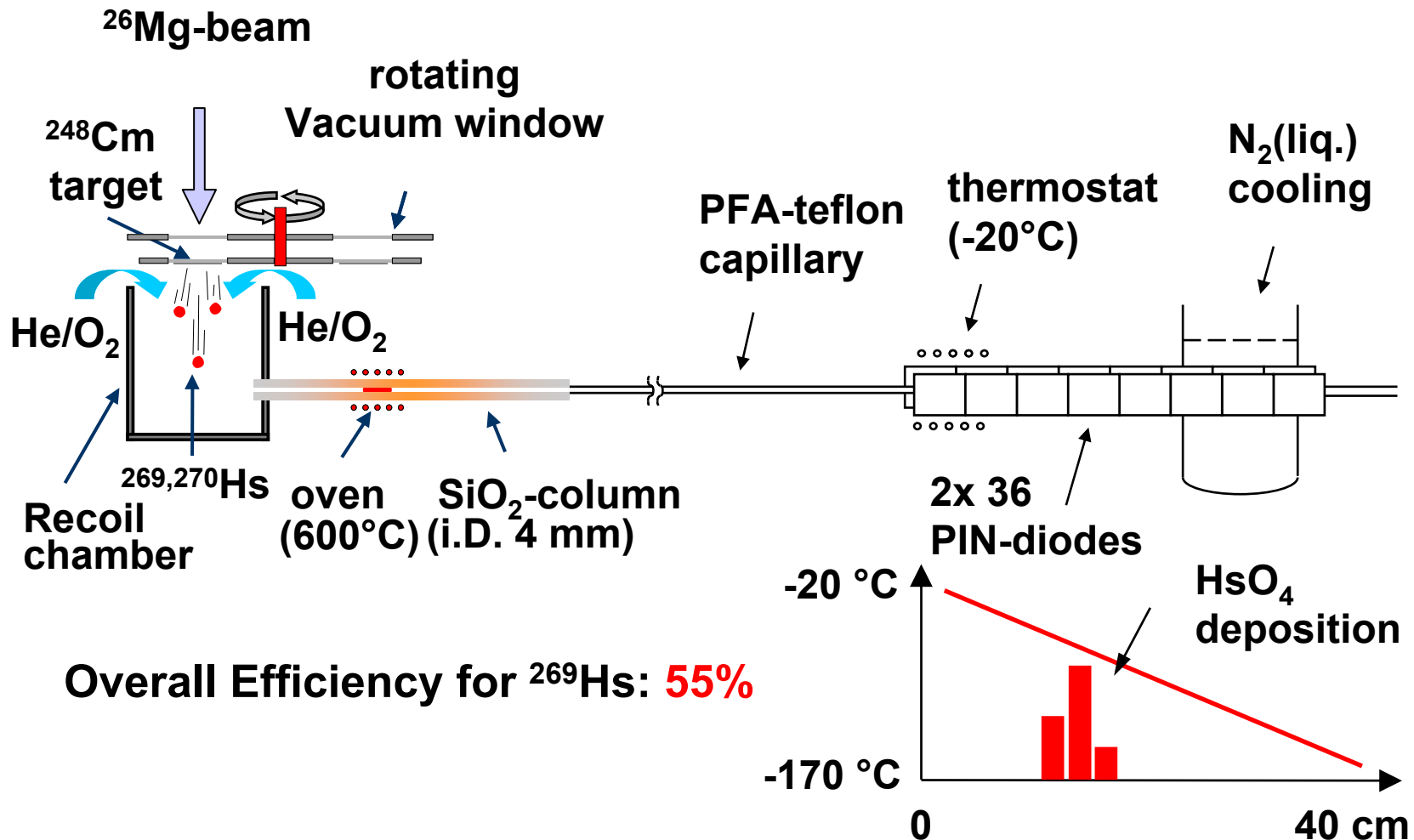
- Named after the German Bundesland Hessen
- Discovered at Gesellschaft für Schwerionenforschung (GSI) 1984
- First chemical investigation: C.E. Düllmann et al., Nature 418, 859 (2002)

# Understanding Superheavy Element Synthesis

Shell correction energies for nuclides with  $Z \geq 82$  and  $N \geq 126$



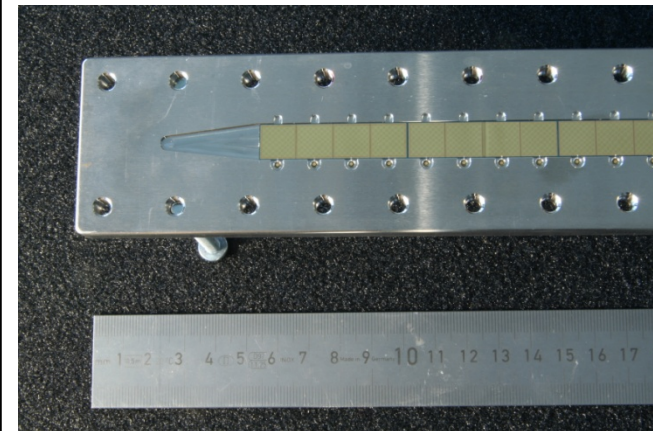
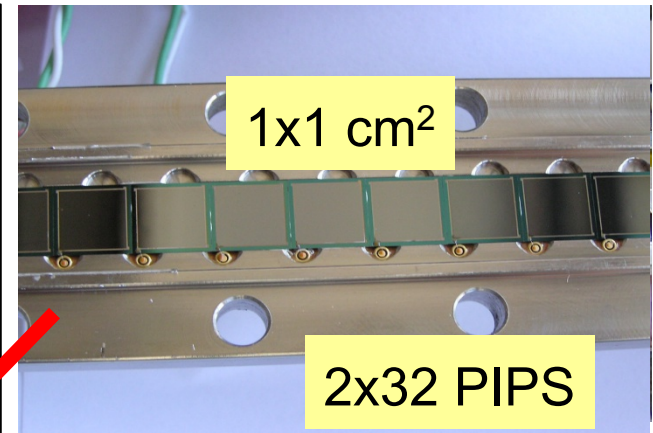
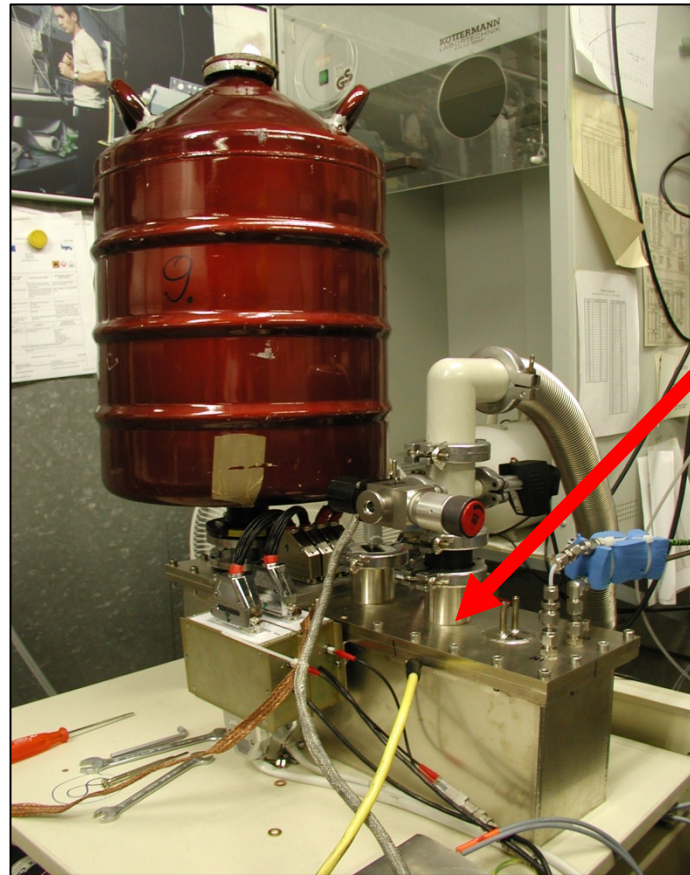
# Chemistry: a highly efficient Hassium separator



# Detection system COMPACT

Version 1: 78% detection eff.

Cryo  
On-line  
Multidetector for  
Physics  
And  
Chemistry of  
Transactinides



Version 2: 93% detection eff.

# Observed Nuclides

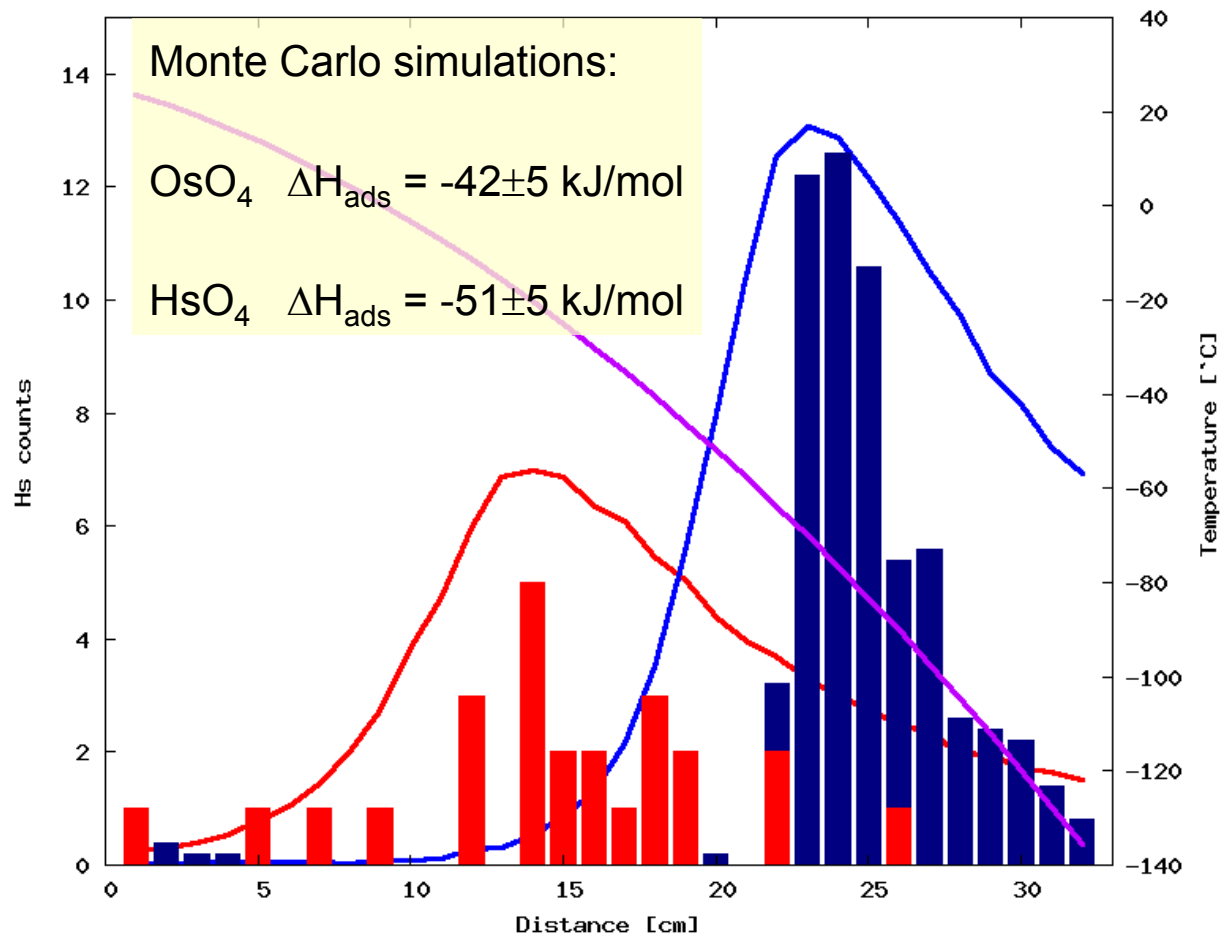
Number of decay chains: 13 7 6

5 new nuclides!

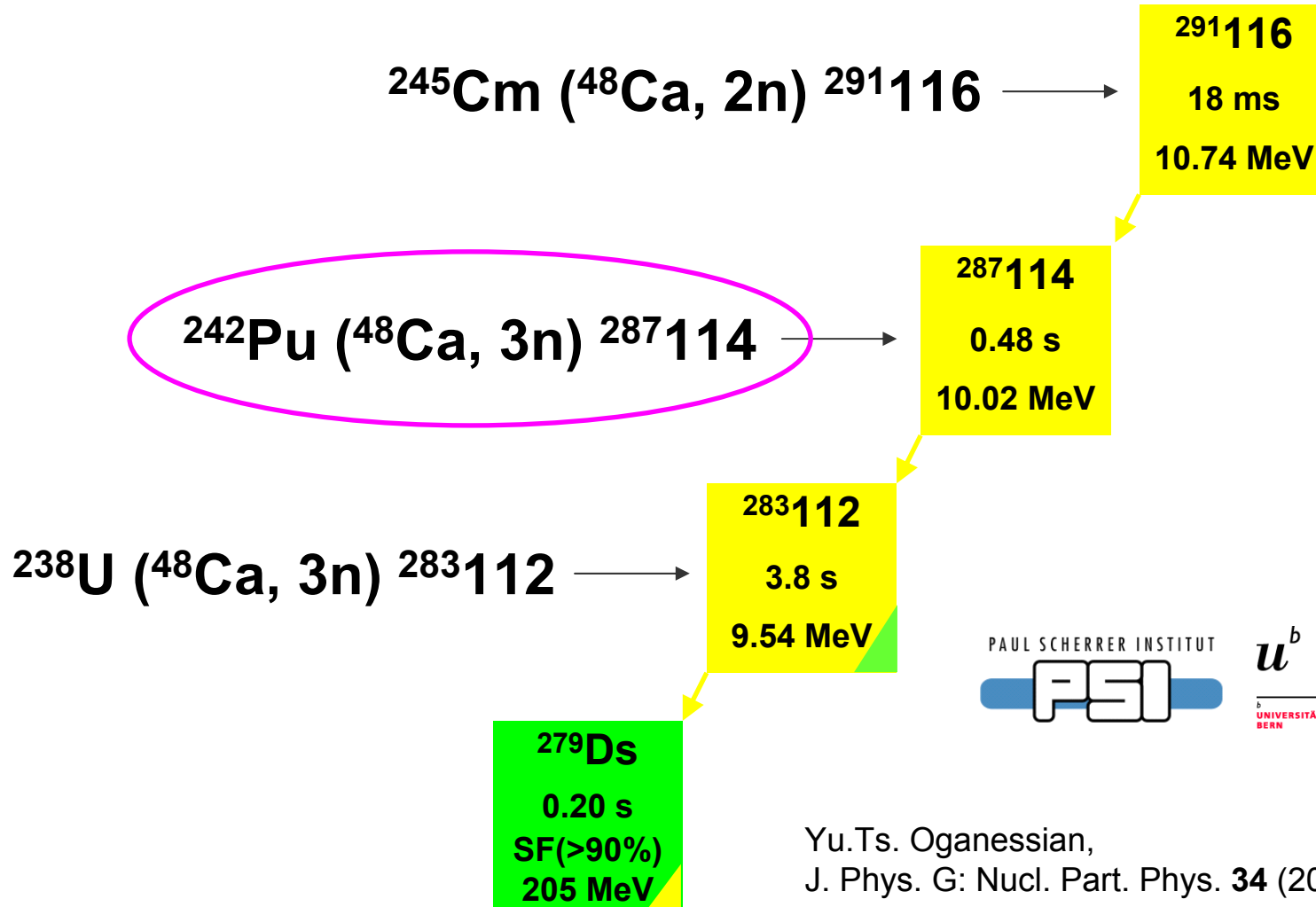
\*  $T_{\alpha}$  is calculated from measured  $Q_{\alpha}$  using  
phenomenological formula from  
A. Parkhomenko, A. Sobiczewski,  
*Acta Physica Polonica* 36, No. 10 (2005)

$^{270}\text{Hs}$ : Jan Dvorak *et al.*, Physical Review Letters **97**, 242501 (2006)  
 $^{271}\text{Hs}$ : Jan Dvorak *et al.*, Physical Review Letters **100**, 132503 (2008)

# Thermochromatography of $\text{HsO}_4$

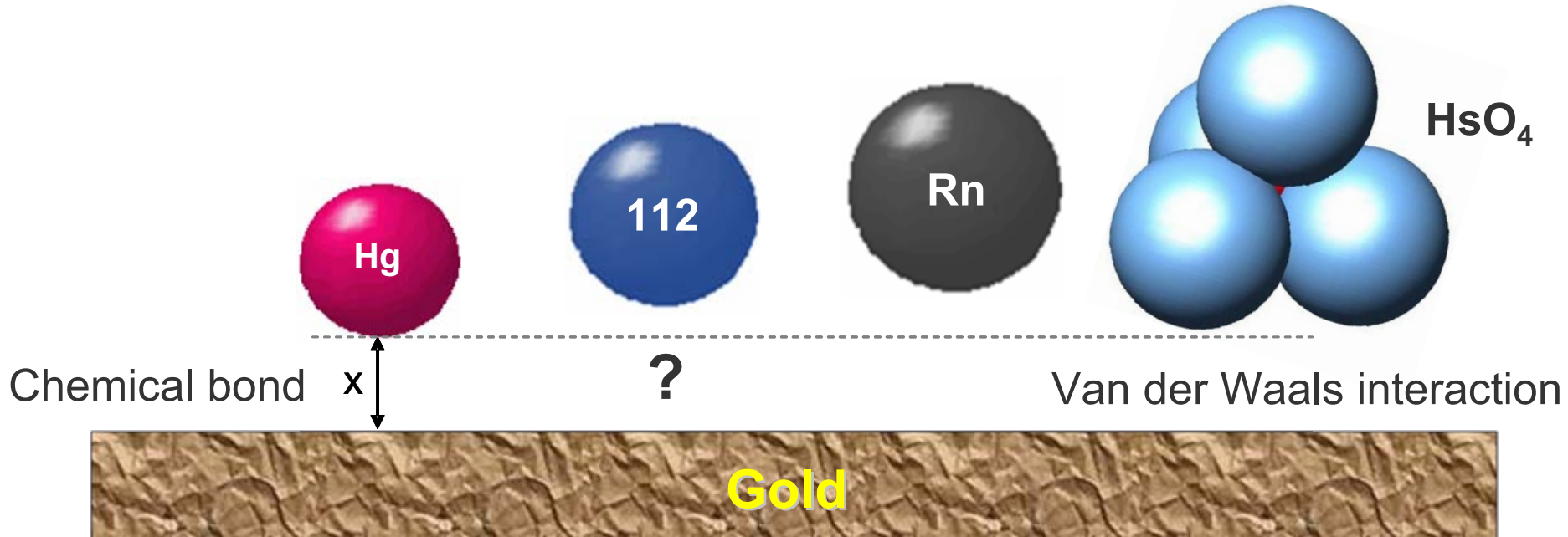


# Element 112 production

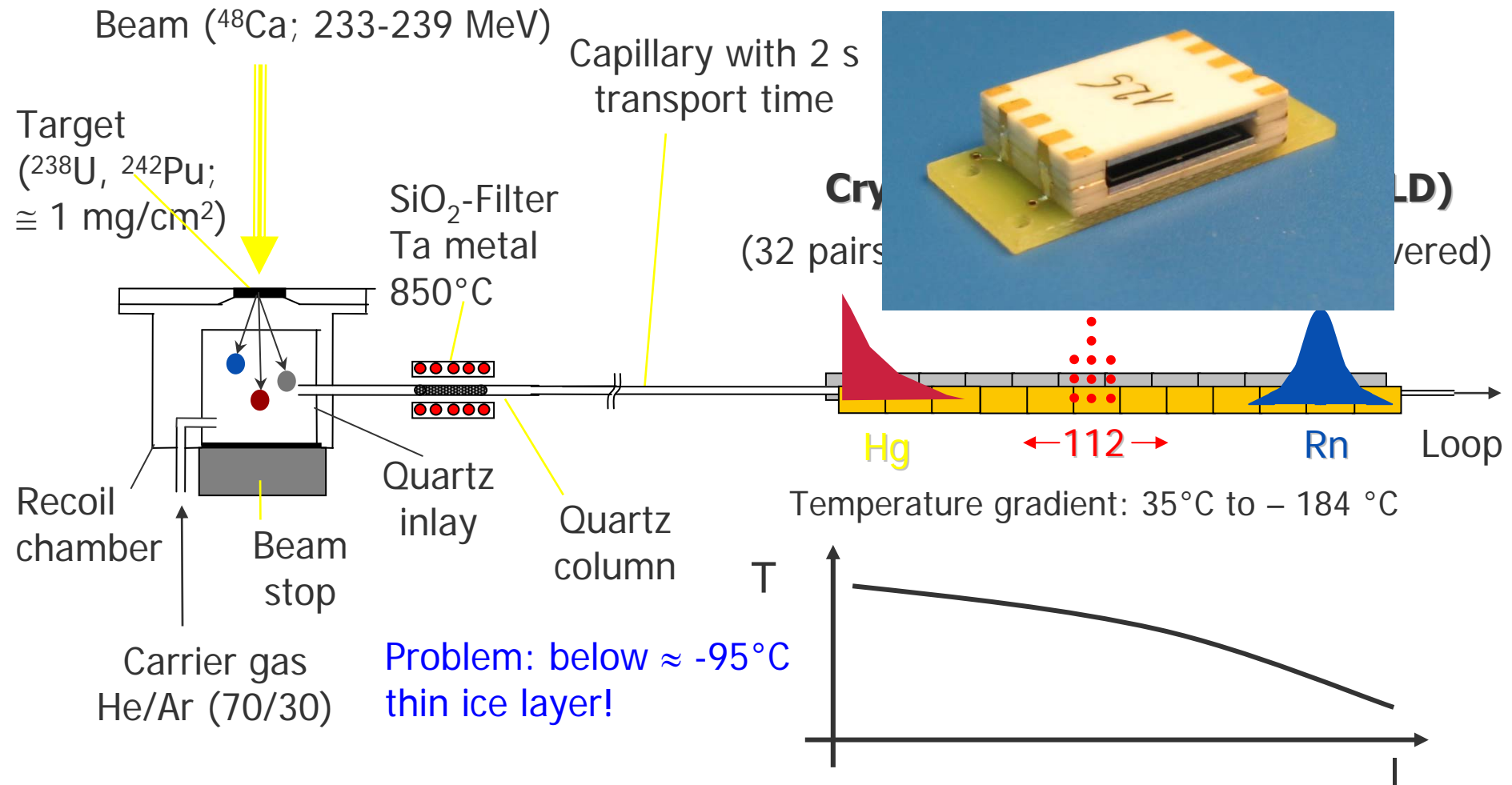


Yu.Ts. Oganessian,  
 J. Phys. G: Nucl. Part. Phys. **34** (2007) R165–R242

# Experimental approach

**Chemisorption****Physisorption**

# The element 112 experiment (IVO Technique)



## NUCLEAR CHEMISTRY

nature

# Panning for ununbium

Andreas Türler

The chemical identification of two atoms of element 112 — scooped from the helium stream they were suspended in using a gold pan — brings the superheavy elements' fabled island of stability into sharper focus.

## LETTERS

## Chemical characterization of element 112

R. Eichler<sup>1,2</sup>, N. V. Aksenov<sup>3</sup>, A. V. Belozerov<sup>3</sup>, G. A. Bozhikov<sup>3</sup>, V. I. Chepigin<sup>3</sup>, S. N. Dmitriev<sup>3</sup>, R. Dressler<sup>1</sup>, H. W. Gäggeler<sup>1,2</sup>, V. A. Gorshkov<sup>3</sup>, F. Haenssler<sup>1,2</sup>, M. G. Itkis<sup>3</sup>, A. Laube<sup>1</sup>, V. Ya. Lebedev<sup>3</sup>, O. N. Malyshev<sup>3</sup>, Yu. Ts. Oganessian<sup>3</sup>, O. V. Petrushkin<sup>3</sup>, D. Piguet<sup>1</sup>, P. Rasmussen<sup>1</sup>, S. V. Shishkin<sup>3</sup>, A. V. Shutov<sup>3</sup>, A. I. Svirikhin<sup>3</sup>, E. E. Tereshatov<sup>3</sup>, G. K. Vostokin<sup>3</sup>, M. Wegrzecki<sup>4</sup> & A. V. Yeremin<sup>3</sup>

Vol 447|3 May 2007|doi:10.1038/nature05761

PAUL SCHERRER INSTITUT  
PSI

 $u^b$ 

UNIVERSITÄT  
BERN

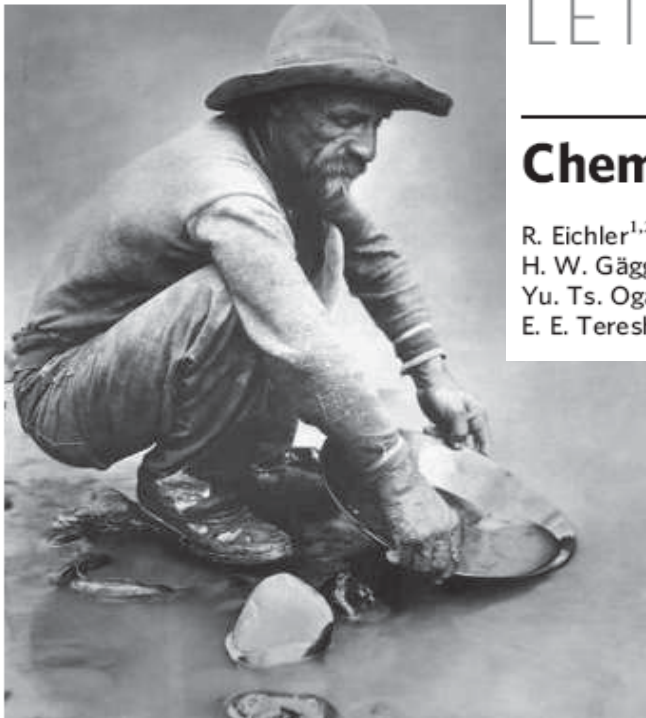
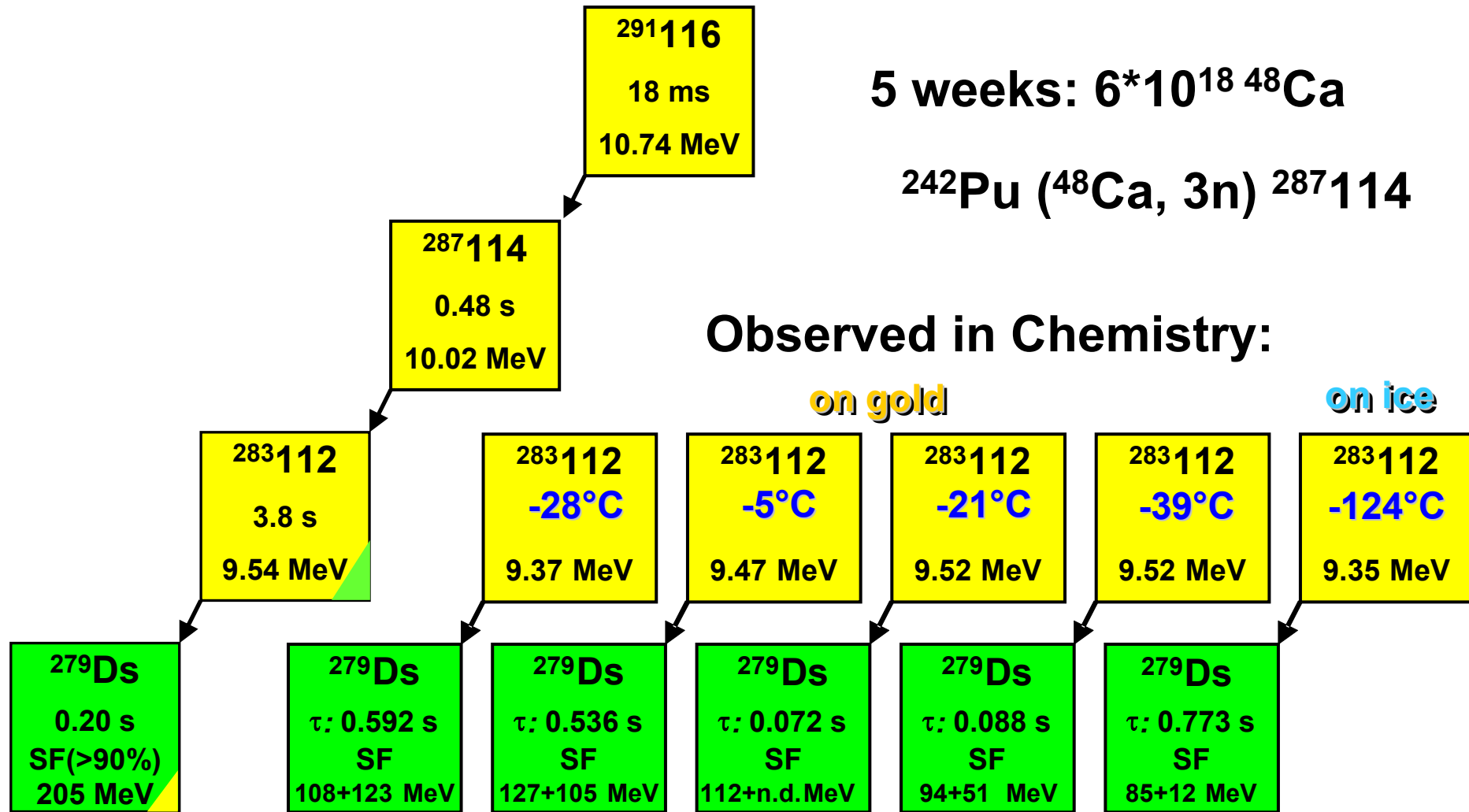


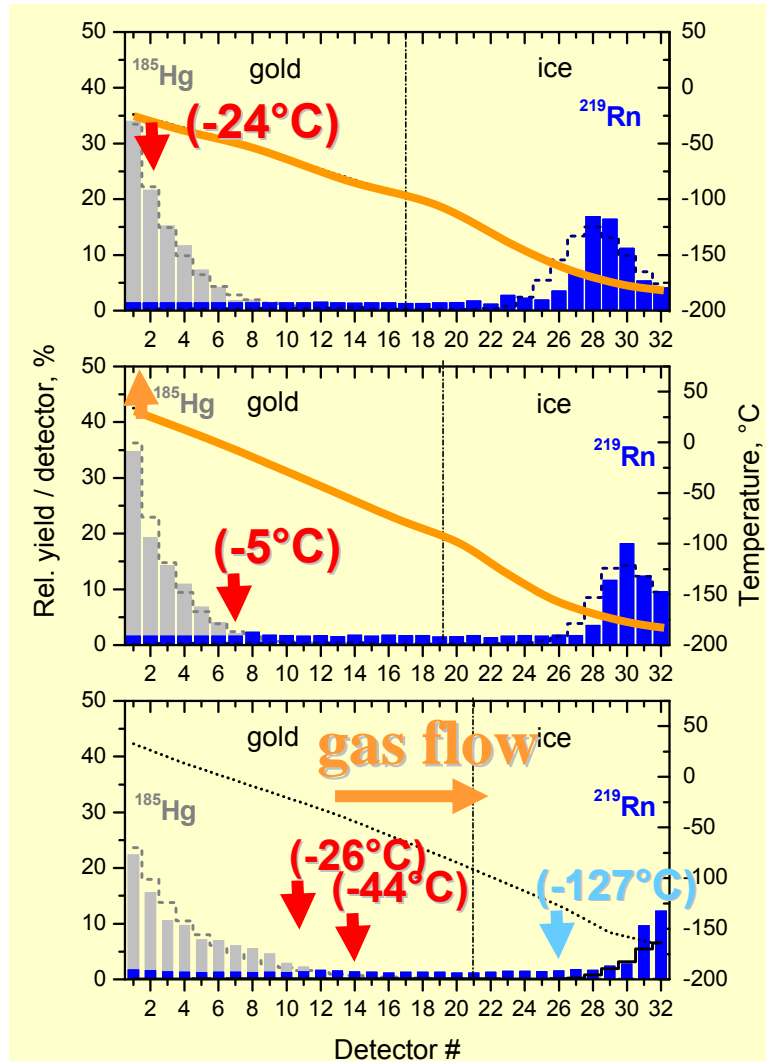
Figure 1 | Element panning, old-school style.

# Observed decay chains: Dubna 2006/2007

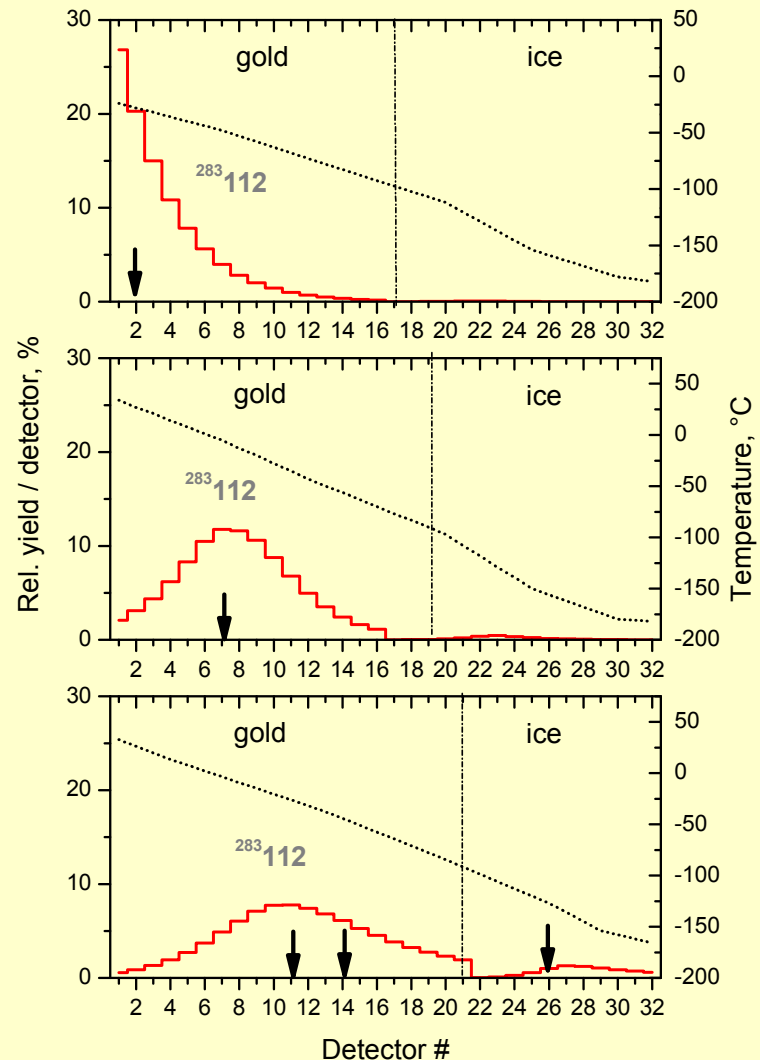


R. Eichler et al. TAN07 and Angew. Chem. Int. Ed. 2008, 47, 3262–3266

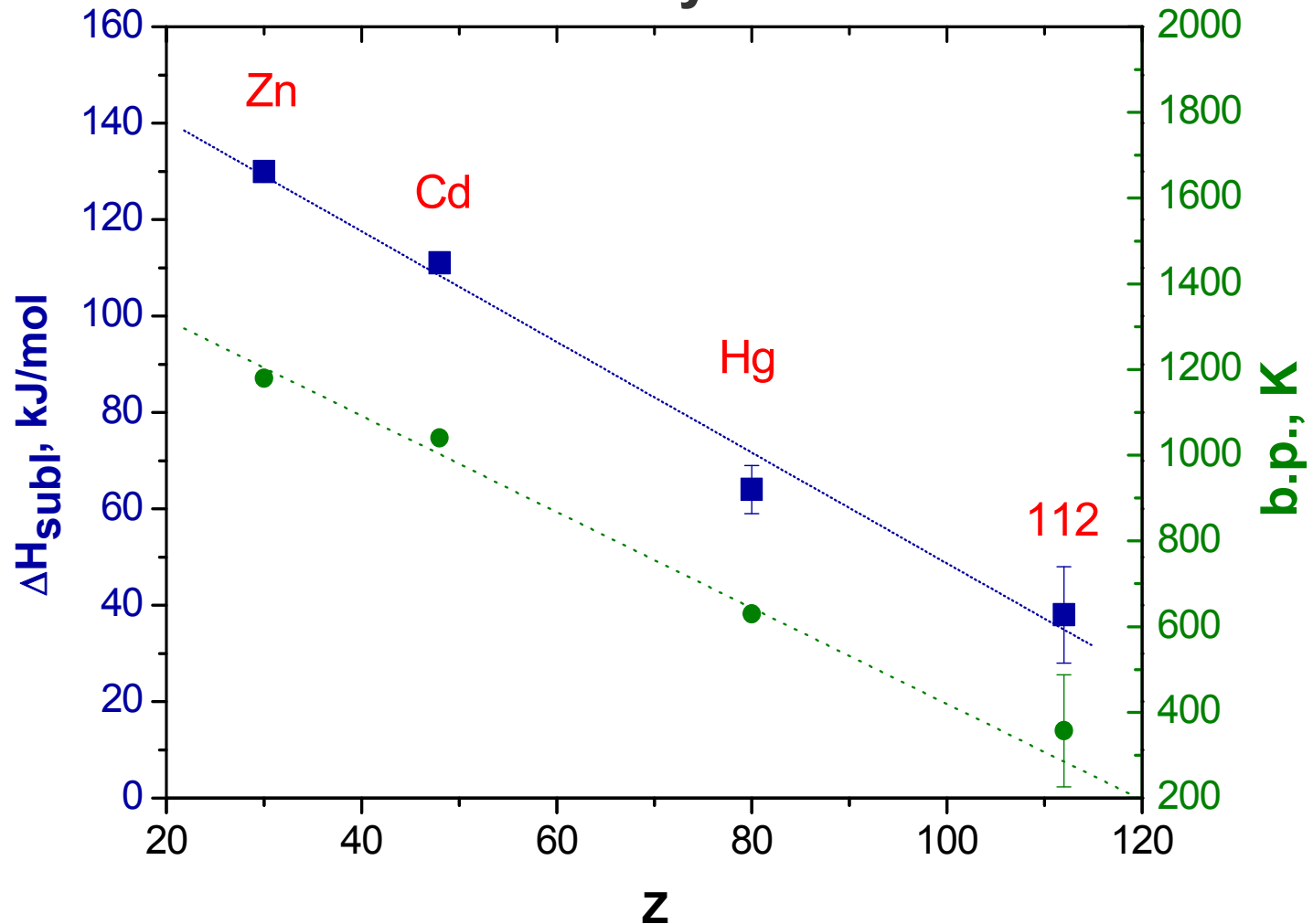
## Experiment



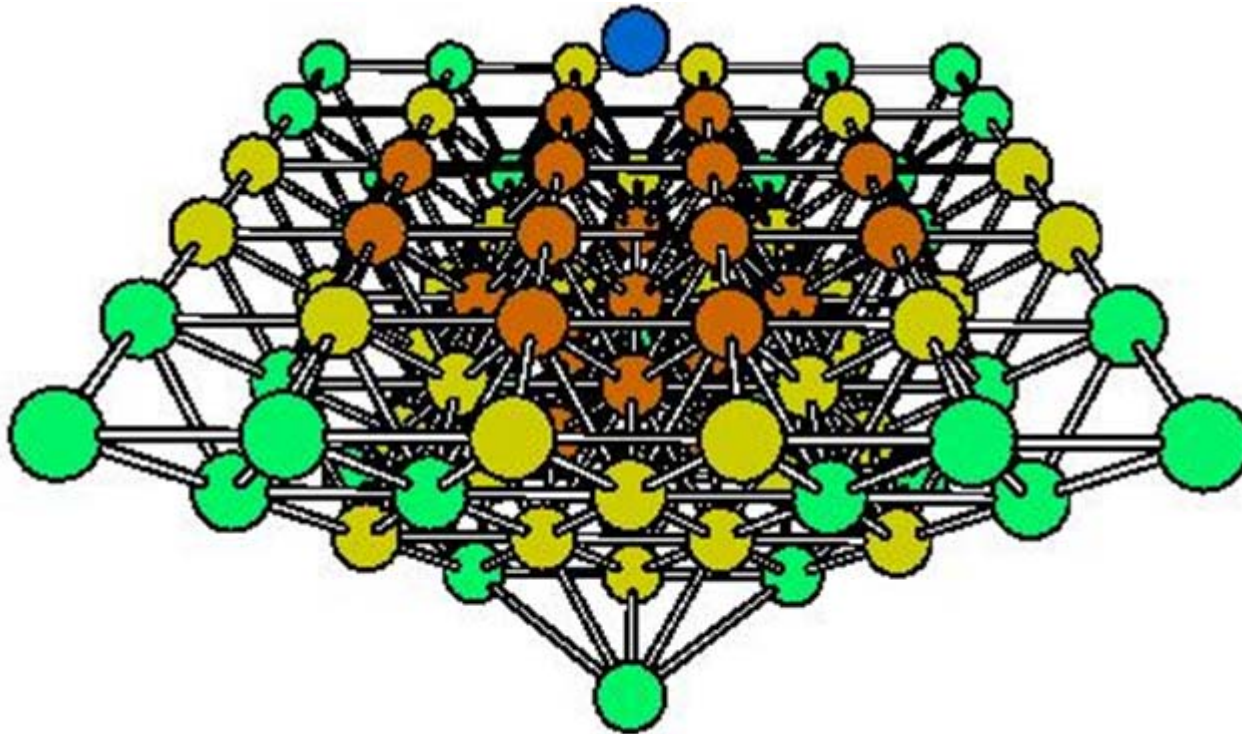
## Model calculation



## From experimental results deduced volatility of element 112



## (E112) ad-atom on the embedded gold cluster



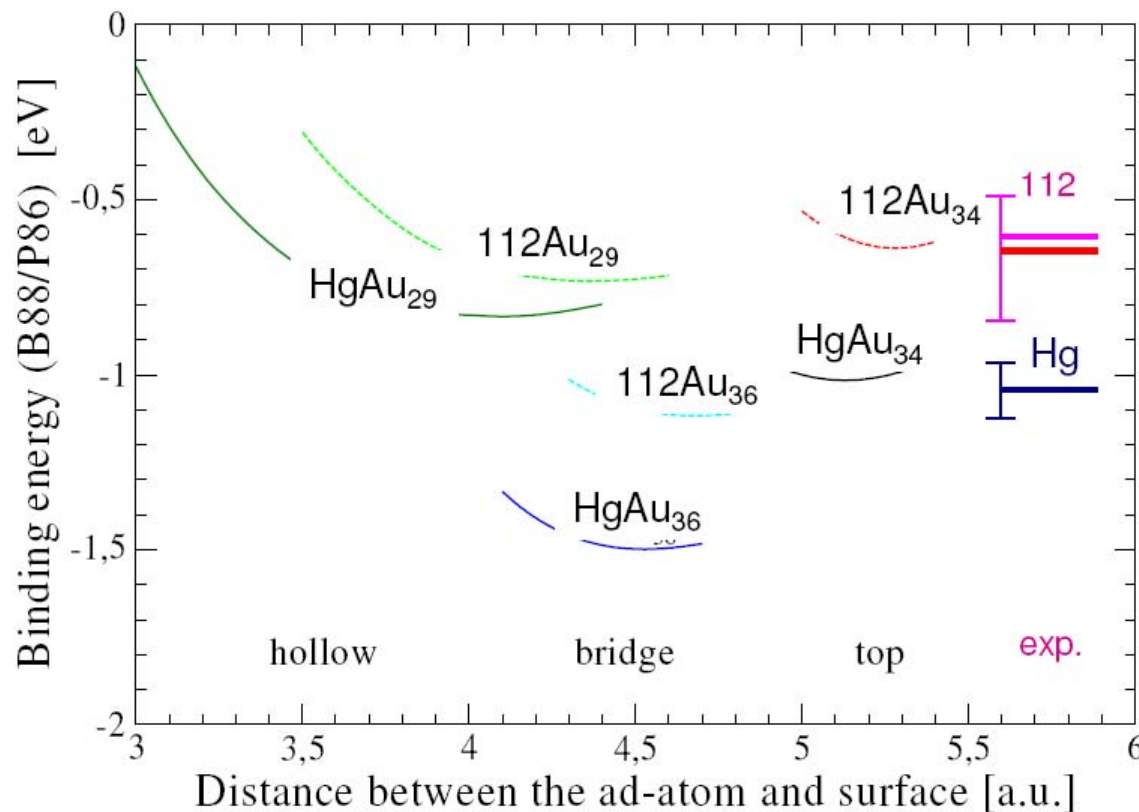
$\approx 35$  Au-cluster atom explicitly treated and  $\approx 100$  Au atoms averaged

C. Sarpe-Tudoran et al., Eur. Phys. J D 24 (2003) 65  
V. Pershina et al., Nucl. Phys. A734 (2004) 200

# Results of Embedded Cluster Calculations

V.G. Pershina et al., contribution to TASCA'08 workshop

112



Predicted:

$T_{\text{ads}}(\text{calc.}) = 0 \text{ }^{\circ}\text{C}$

$\Delta H_{\text{ads}}(\text{calc.}) = -62 \text{ kJ/mol}$

[V. Pershina *et al.*,  
Nucl. Phys. A **734**, 200 (2004);  
*ibid* **787**, 381 (2007) ]

Observed:

$T_{\text{ads}}(\text{exp.}) = -5 \text{ }^{\circ}\text{C}$

$\Delta H_{\text{ads}}(\text{exp.}) = -52_{-4}^{+20} \text{ kJ/mol}$

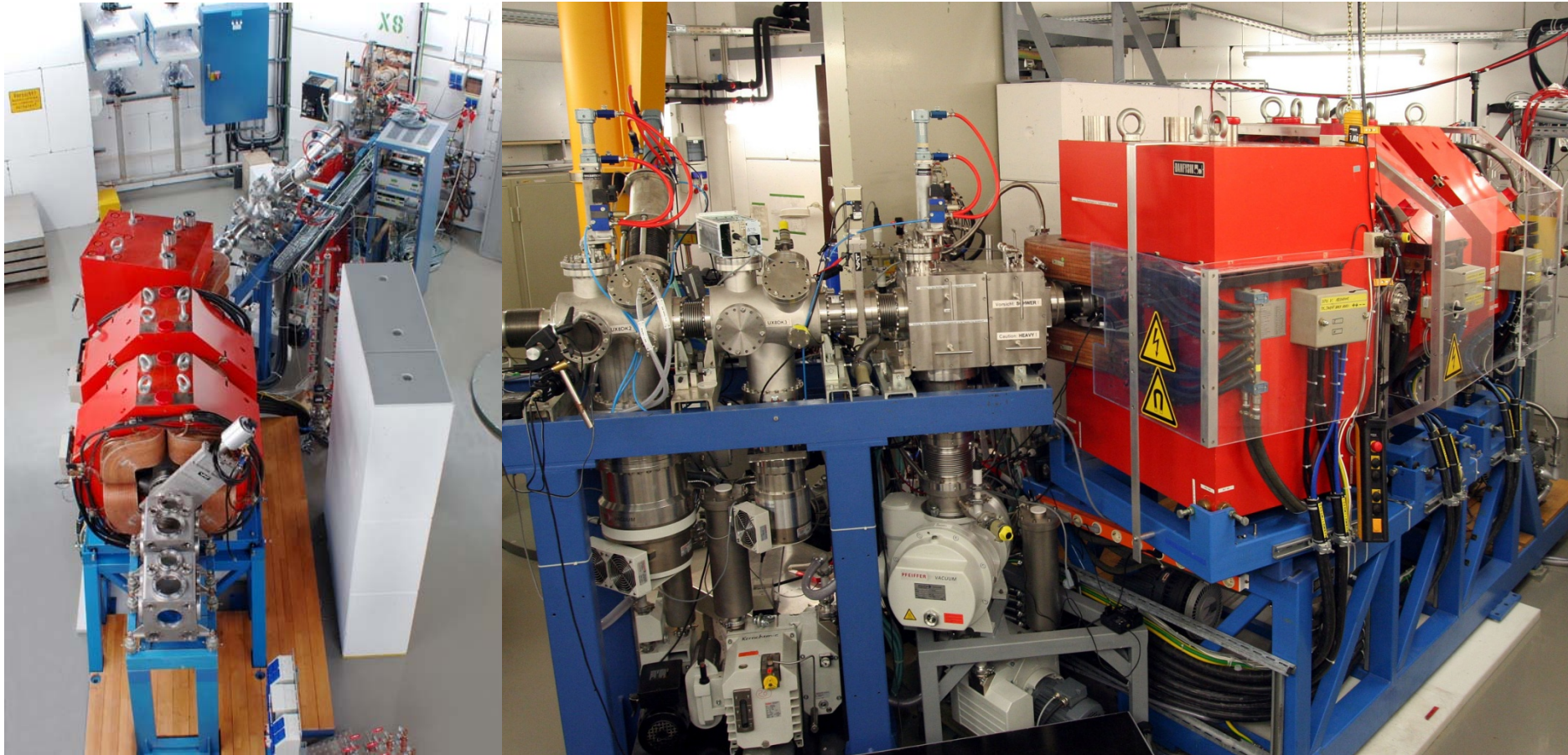
[R. Eichler, *et al.*  
Nature, **447**, 72 (2007)]

C. Sarpe-Tudoran *et al.* J. Chem. Phys. **126**, 174702 (2007)]

# Future of transactinide chemistry: TASCA



# TASCA - Trans Actinide Separator and Chemistry Apparatus



TASCA home page: <http://www-w2k.gsi.de/tasca/>

# TASCA collaboration



magnets, power supplies, vacuum chambers and pumps, construction of cave and shielding, beam line, crane, electronics, €'s



magnetic field maps, ion optics, Monte-Carlo simulations, control system, differential pumping system, vacuum windows, focal plane detector & electronics, €'s



targets, recoil chamber HTM, €'s



recoil chamber SIM, €'s



Monte Carlo simulations, ion optics, TOF



Monte Carlo simulations, ion optics, electronics

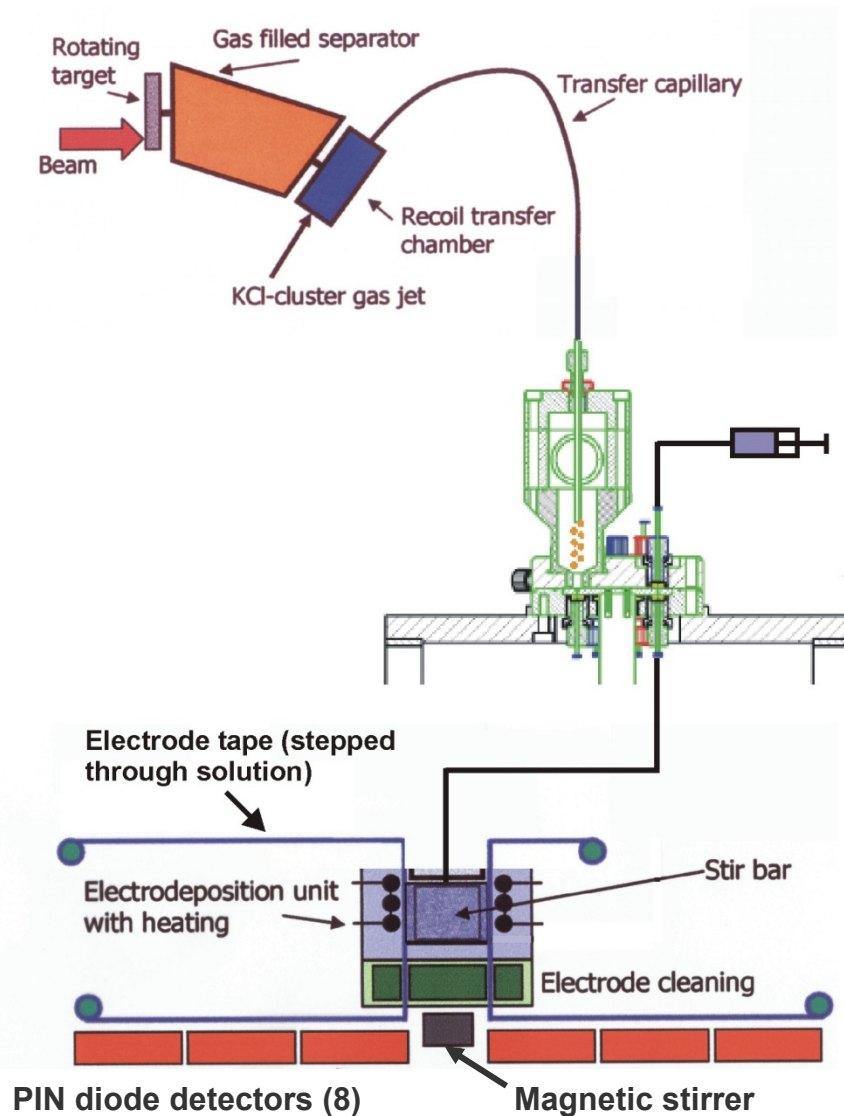


differential pumping system, ion optics, focal plane detector

# TASCA vacuum chambers

M)

optical



## An example of a future electrodeposition experiment with Hassium

Coupling scheme of

TASCA – RTC – ALOHA –  
ELECTRODEPOSITION with  
two electrode tapes and two  
detector arrays

**The End!**  
**(of the talk)**  
**Thank you for listening**